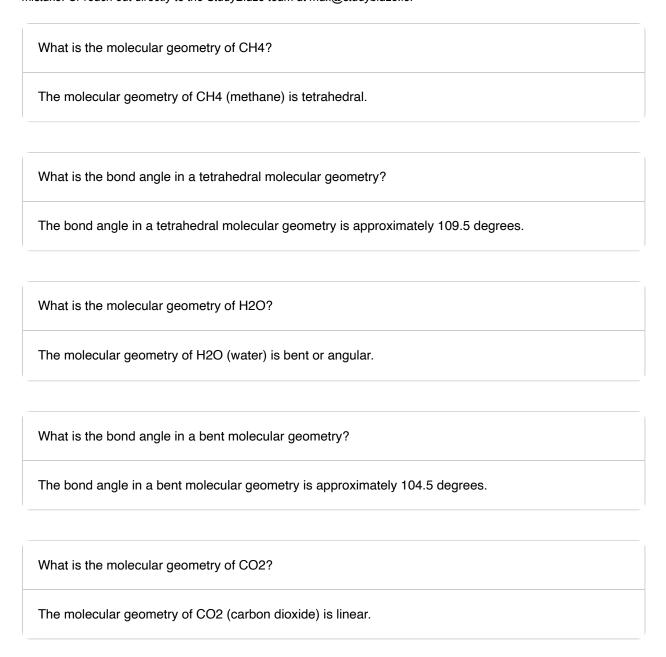


Molecular Geometry Flashcards PDF

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What is the bond angle in a linear molecular geometry?
The bond angle in a linear molecular geometry is 180 degrees.
What is the molecular geometry of NH3?
The molecular geometry of NH3 (ammonia) is trigonal pyramidal.
What is the bond angle in a trigonal pyramidal molecular geometry?
The bond angle in a trigonal pyramidal molecular geometry is approximately 107 degrees.
What is the molecular geometry of BF3?
The molecular geometry of BF3 (boron trifluoride) is trigonal planar.
What is the bond angle in a trigonal planar molecular geometry?
The bond angle in a trigonal planar molecular geometry is 120 degrees.