

Flashcards For Organic Chemistry Reactions PDF

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What is the general reaction type for the conversion of an alkene to an alkane?
Hydrogenation.
What type of reaction involves the addition of a nucleophile to a carbonyl compound?
Nucleophilic addition.
What is the product of the reaction between an alcohol and a carboxylic acid?
An ester.
What type of reaction is characterized by the loss of a small molecule, such as water, during the formation of a larger molecule?
Condensation reaction.
What is the mechanism called when a halogen adds across a double bond?
Electrophilic addition.



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What is the primary product of the oxidation of a primary alcohol?
An aldehyde.
What type of reaction occurs when an alkyl halide reacts with a strong base to form an alkene?
Elimination reaction.
What is the name of the reaction that converts an alkene to a diol using osmium tetroxide?
Syn-dihydroxylation.
What is the result of the reaction between an amine and a carbonyl compound?
An imine.
What type of reaction involves the transfer of electrons from one species to another?
Redox reaction.