

Determining Density Worksheet

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Part 1: Building a Foundation
What is the formula for calculating density?
Hint: Think about the relationship between mass and volume.
O Density = Volume / Mass
O Density = Mass / Volume
O Density = Mass x Volume
O Density = Volume x Mass
Which of the following are units of density? (Select all that apply)
Hint: Consider common units used in science.
☐ g/cm³
☐ kg/m³
\square m/s ²
Define density in your own words.
Hint: Think about how mass and volume relate to each other.
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List two factors that can affect the density of a substance.



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Hint: Consider physical properties that might change density.
1. Factor 1
2. Factor 2
Which of the following statements is true about density?
Hint: Think about the properties of different substances.
Opensity is independent of temperature.
O Density is the same for all substances.
O Density can help determine if an object will float or sink.
Opensity is measured in meters.
Part 2: Comprehension and Application
Why does ice float on water?
Why does ice float on water? Hint: Consider the densities of ice and water.
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Explain how the concept of density is used to determine whether an object will float or sink in a

fluid.



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Hint: Consider the relationship between the object's density and the fluid's density.
A block of wood has a mass of 200 grams and a volume of 250 cm ³ . What is its density?
Hint: Use the formula for density to calculate.
○ 0.8 g/cm³
○ 1.25 g/cm³
2.5 g/cm ³
○ 0.5 g/cm³
You have a piece of metal with a density of 7.8 g/cm ³ . Which of the following will happen if you place it in water (density = 1 g/cm ³)? (Select all that apply)
Hint: Consider the relationship between the densities.
☐ It will float.
☐ It will sink.
It will remain suspended.
☐ It will dissolve.
Describe a real-world scenario where understanding the density of a material is crucial for safety or efficiency.
Hint: Think about industries where density plays a key role.
Doub O. Analysis - Evaluation and Overtica
Part 3: Analysis, Evaluation, and Creation

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If the temperature of a gas increases, what is likely to happen to its density?
Hint: Consider how temperature affects gas behavior.
Opensity will increase.
Opensity will decrease.
Opensity will remain the same.
Opensity will double.
Consider two liquids, A and B. Liquid A has a density of 0.9 g/cm³, and Liquid B has a density of 1.1 g/cm³. Which of the following statements are true? (Select all that apply)
Hint: Think about how the densities compare.
Liquid A will float on Liquid B.
Liquid B will float on Liquid A.
☐ Both liquids will mix evenly.
Liquid A is denser than Liquid B.
Analyze how pressure changes might affect the density of a gas in a closed container.
Hint: Consider the relationship between pressure and volume.
Which method would be most effective for measuring the volume of an irregularly shaped object?
Hint: Think about techniques used in science labs.
○ Using a ruler to measure dimensions
○ Using a balance scale
○ Water displacement method
Calculating based on mass
Evaluate the following scenarios and determine which would require consideration of density. (Select all that apply)

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Hint: Think about engineering and design applications.



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Designinga ship	
Painting a house	
Building a bridge	
Cooking pasta	
opose a method to determine the density of a new material that is both solid and porous. Explaises steps and reasoning behind your method.	n
t: Consider how to measure mass and volume accurately.	
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