

Radioactivity Quiz PDF

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Which of the following isotopes are used in nuclear reactors?
☐ Uranium-235
☐ Plutonium-239
Carbon-14
☐ Thorium-232
Which type of radiation consists of helium nuclei?
○ Alpha particles
○ Beta particles
○ Gamma rays
○ Neutrons
Which of the following materials is most effective at blocking gamma rays?
○ Paper
○ Aluminum
○ Lead
○ Water
Describe the role of radioactivity in medical diagnostics and treatment, providing specific examples.

What is the main application of gamma rays in medicine?



○ Diagnostic imaging○ Sterilization	
Cancer treatment	
Bone density measurement	
Describe the differences between alpha, beta, penetration power.	and gamma radiation in terms of composition and
	/2
Discuss the historical significance of Marie C	urie's contributions to the field of radioactivity.
	/1
What are the environmental impacts of nuclea	er accidents, and how can they be mitigated?
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How does the principle of radioactive dating work, and what are its applications?



What is the process by which unstable atomic nuclei lose energy by emitting radiation called?	
○ Fusion	
○ Fission	
○ Radioactivity	
Olonization	
Explain the concept of half-life and its significance in radioactive decay.	
	_//
Which nuclear reactions release energy?	
☐ Fission	
☐ Fusion	
☐ Ionization	
☐ Combustions	
Which of the following are applications of radioactivity in industry?	
☐ Food irradiation	
☐ Space exploration	
☐ Water purification	

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What is the primary purpose of a Geiger-Müller counter?



○ To measure temperature
○ To detect ionizing radiation
○ To calculate half-life
○ To enrich uranium
What are potential health effects of ionizing radiation exposure?
Burns
☐ Radiation sickness
☐ Increased cancer risk
☐ Enhanced immune function
What are the primary safety measures to minimize radiation exposure?
☐ Time
☐ Distance
Shieldin
☐ Temperature control
Which of the following are types of ionizing radiation?
Which of the following are types of ionizing radiation? Alpha particles
☐ Alpha particles
☐ Alpha particles ☐ Beta particles
☐ Alpha particles☐ Beta particles☐ Gamma rays
☐ Alpha particles☐ Beta particles☐ Gamma rays
□ Alpha particles□ Beta particles□ Gamma rays□ Ultraviolet light
☐ Alpha particles ☐ Beta particles ☐ Gamma rays ☐ Ultraviolet light Who discovered natural radioactivity?
□ Alpha particles □ Beta particles □ Gamma rays □ Ultraviolet light Who discovered natural radioactivity? □ Albert Einstein
□ Alpha particles □ Beta particles □ Gamma rays □ Ultraviolet light Who discovered natural radioactivity? □ Albert Einstein □ Henri Becquerel
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 □ Alpha particles □ Beta particles □ Gamma rays □ Ultraviolet light Who discovered natural radioactivity? □ Albert Einstein □ Henri Becquerel □ Marie Curie □ Niels Bohr What is the SI unit of radioactivity? □ Sievert

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Which radioactive isotope is commonly used in radiocarbon dating?	
○ Uranium-235	
○ Carbon-14	
○ Radon-222	
○ Thorium-232	