

Heat and Temperature Quiz PDF

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Describe how the second law of thermodynamics applies to heat engines.	
	//
What is the significance of the latent heat of vaporization in phase changes?	
	//
low does thermal expansion affect the design of bridges?	
	//

Why is water's high specific heat capacity important for climate regulation?



Describe an experiment to measure the specific heat capacity of a metal.
Which factors affect the rate of heat transfer by conduction? (Select all that apply)
☐ Temperature difference
☐ Surface area
Material thickness
Color of the material
Which temperature scale is based on the absolute zero point?
○ Celsius
○ Fahrenheit
○ Kelvin
○ Rankine
Which of the following are methods of heat transfer? (Select all that apply)
☐ Conduction
☐ Convection
Radiation
Evaporation
What does temperature measure in a substance?

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○ Total energy
Average kinetic energy of particles
O Potential energy
○ Chemical energy
Which of the following is a measure of heat energy required to change the temperature of substance?
○ Thermal conductivity
○ Specific heat capacity
○ Thermal expansion
○ Heat flux
What is the process called when a liquid turns into a gas?
FreezingCondensation
○ Vaporization
Sublimation
Which method of heat transfer involves the movement of fluids?
○ Conduction
○ Convection
Radiation
○ Insulation
Which law of thermodynamics states that energy cannot be created or destroyed?
○ Zeroth Law
○ First Law
○ Second Law
○ Third Law
What happens to a metal rod when it is heated?
○ It contracts
O It expands
O It remains the same
O It melts

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Which of the following are examples of phase changes? (Select all that apply)
☐ Melting
☐ Boiling
☐ Freezing
☐ Sublimation
Explain the difference between heat and temperature.
Which statements are true about thermal equilibrium? (Select all that apply)
☐ It occurs when two objects reach the same temperature.
Heat continues to flow between objects.
No net heat flow occurs between objects.
☐ It can only occur in solids.
Which of the following are units of temperature? (Select all that apply)
☐ Celsius
☐ Kelvin
☐ Joule
☐ Fahrenheit
What is the primary unit of measurement for heat in the International System of Units (SI)?
○ Calorie
O Joule
○ Fahrenheit
○ Kelvin

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Which of the following statements about specific heat capacity are correct? (Select all that apply)



☐ It is the same for all substances.
☐ It determines how much heat is needed to change the temperature of a substance.
☐ Water has a high specific heat capacity.
☐ It is measured in Joules per kilogram per degree Celsius.