

Beta Decay Quiz PDF

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What is emitted during beta-minus decay?
ProtonElectronPositronNeutron
Which type of beta decay involves the conversion of a neutron into a proton?
Alpha decayBeta-plus decayGamma decayBeta-minus decay
Which of the following particles is nearly massless and emitted during beta decay?
PhotonElectronProtonNeutrino
Describe how beta-plus decay affects the atomic number and mass number of an element.

Which conservation laws are applicable to beta decay? (Select all that apply)



Conservation of charge	
Conservation of baryon number	
Conservation of mass-energy	
Conservation of lepton number	
Which of the following are emitted during beta-plus decay? (Select all that apply)	
□ Positron	
□ Neutrino	
☐ Antineutrino	
☐ Electron	
How does beta decay contribute to the stability of isotopes in nature?	
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Which of the following statements about beta decay are true? (Select all that apply)	
☐ It changes the element's identity.	
☐ It involves a change in atomic number.	
☐ It is a form of nuclear fission.	
☐ It emits gamma rays.	
Explain the historical context of the discovery of beta decay and its impact on nuclear physics.	
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Which conservation law is not directly involved in beta decay?



 Conservation of charge Conservation of mass-energy Conservation of angular momentum Conservation of momentum
In beta-plus decay, what particle is emitted from the nucleus?
○ Electron ○ Proton
O Positron
○ Neutron
What is the charge of a beta particle emitted during beta-minus decay?
○ Positive
Neutral
Double positiveNegative
What happens to the atomic number of an element undergoing beta-minus decay?
○ Increases by 1
○ Remains the same
Opubles
O Decreases by 1
What role do neutrinos play in beta decay, and why are they important for conservation laws?

Discuss the significance of beta decay in medical applications, providing at least one example.



Which particles are involved in the process of beta-minus decay? (Select all that apply)	
☐ Proton	
☐ Electron	
☐ Antineutrino	
☐ Neutron	
What is the primary purpose of beta decay in nuclear physics?	
○ To increase atomic mass	
○ To decrease atomic number	
○ To produce gamma rays	
○ To stabilize an unstable nucleus	
Data descriptions which of the following product properties? (Coloct all that emply)	
Beta decay affects which of the following nuclear properties? (Select all that apply)	
Atomic number	
Charge	
Element identity	
☐ Mass number	
In which applications is beta decay utilized? (Select all that apply)	
Carbon dating	
☐ Metal refining	
☐ Nuclear power generation	
Explain the process of beta-minus decay, including the particles involved and the changes in	the

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nucleus.



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